

## ELEGANT, RAPID AND SIMPLE PROCEDURE FOR LABELING OF ANTIBODIES WITH FLUORESCENT ZnCd/S QUANTUM DOTS FOR LASER ABLATION ICP-MS DETECTION

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### Abstract

Antibody labeling with a signal-providing tag (e.g. an enzyme, chromophore, fluorophore, metal chelate, radioisotope, etc.) is a key procedure in bioanalytical techniques such as an enzyme-linked immunosorbent assay, western blot, or immunohistochemical analysis. The process may be laborious, impair the activity, and time-consuming. Concisely, in this work, an elegant method of rapid and effective antibody labeling by quantum dots was developed taking advantage of partial reduction of antibody structure and UV-induced quantum dot formation. Zinc and cadmium with the thiol group of the antibody as capping agents stabilizing the fluorescent nanoparticle while the biorecognition capabilities were maintained. The ZnCd/S quantum dots creation has to prosper detected with capillary electrophoresis (CE) and laser ablation ICP-MS (LA-ICP-MS).

**Keywords:** Antibody, antigen, quantum dots, cleavage disulfide bonds, fluorescent

### 1. INTRODUCTION

Antibody labeling techniques are extremely an important tool for protein detection and it can be done by direct or indirect labeling. The disadvantage of the process may impair antibody activity with excessive labeling, steric hindrance, and alter activity for high molecular weight such as enzymes, discontinuity conjugation for multiple fluorophores and enzymes, laborious and time-consuming [1,2].

Quantum dots (QDs) are described as the most popular fluorescent inorganic nanoparticles (FINPs) which have properties such as good biocompatibility, stable photoluminescence, and narrow emission band. These materials attract attention in recent years and play a key role in a different field such as ion detection, bio-labeling or bioimaging [3,4]. Antibodies are widely used as targeting moieties with QDs for specific cell labeling. They interact with the host cell and remain adhered to the surface or internalized by endocytosis [5]. The labeled antibody with quantum dots has been detected with laser ablation ICP-MS (LA-ICP-MS) for trace element determination [6] and capillary electrophoresis to verified the formation of QDs-antibody [7].

In this research work, for the first time, ZnCd/S quantum dots prepared by UV-induced synthesis with partially reduced antibody as a capping agent were used for immunoanalysis with elemental analysis. UV irradiation duration and metal ratio were optimized to achieve a high fluorescent intensity. To explore ZnCd/S formation, LA-ICP-MS, and CE were used as superior instruments for quantum dots characterization.

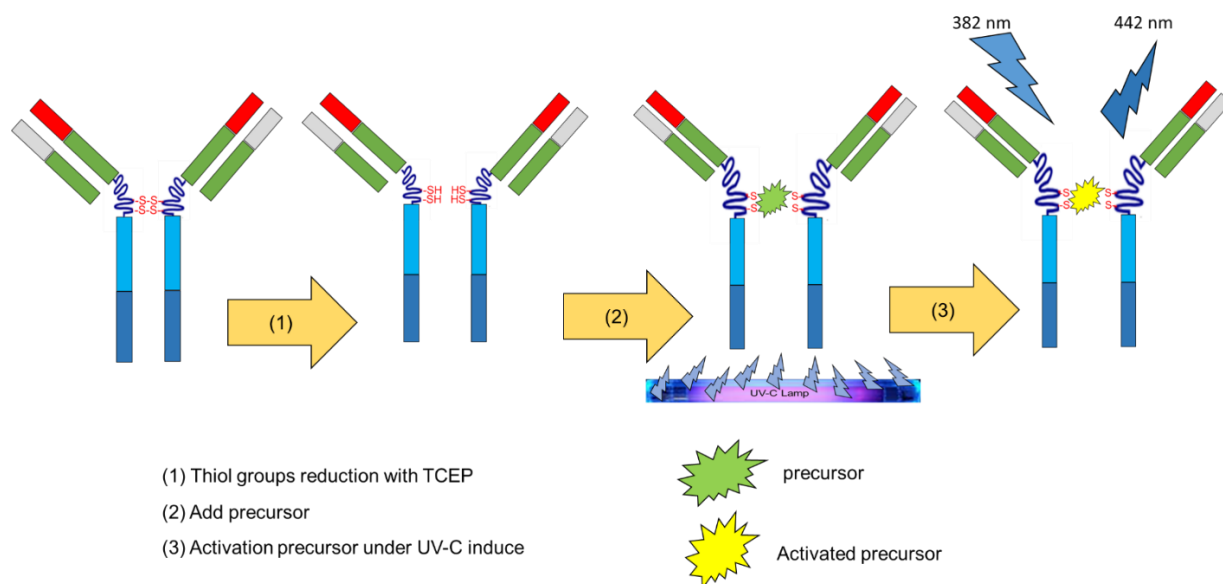
## 2. MATERIALS AND METHODS

### 2.1. Materials

Anti-mouse IgG H&L (ab6708) was purchased from Abcam (Cambridge, UK). All chemicals and solutions used in this study provisioned daily without further purification were purchased from Sigma-Aldrich (St. Louis, MO, USA) in ACS purity. Millipore water was used throughout the experiment.

### 2.2. Simultaneous QDs Synthesis and Antibody Conjugation Process

Quantum dots-antibody (QDs-Ab) probe was prepared according to the simple *modus operandi*. In order to reduction of 1 mg/mL of the anti-mouse IgG, tris(2-carboxyethyl) phosphine (TCEP) was used in the concentration of 25 mM at 37 °C for 90 min with sodium phosphate buffer ( $\text{Na}_2\text{HPO}_4/\text{NaH}_2\text{PO}_4$ ) pH=7 in a total volume of 10  $\mu\text{L}$ . Precursors containing zinc acetate ( $4 \times 10^{-5}$  mM) and cadmium acetate ( $6 \times 10^{-6}$  mM) were mixed with reduced anti-mouse IgG in 100  $\mu\text{L}$  sodium phosphate buffer pH=7. The prepared solution was pipetted into the UV-transparent 96-well plate (Corning, USA) and placed into the UV transilluminator (Vilber Lourmat, Marne-la-Vallee Cedex, France) with wavelength of 254 nm. Resulting solution was analyzed by fluorescence spectrometry. Fluorescent intensity was measured in the range of 230 to 700 nm with Synergy H1 (BioTek, USA). **Scheme 1** is illustrating the overall procedure.



**Scheme 1** Anti-mouse IgG reduction and creation of ZnCd/S quantum dots with antibody conjugation

### 2.3. Capillary electrophoresis (CE)

Prepared QDs-Ab was analyzed by CE instrument 7100 (Agilent Technologies, Germany) with absorbance detection at a wavelength of 254 nm. Fused silica capillary with an internal diameter of 75  $\mu\text{m}$ , with a total length of 64.5 cm, and an effective length of 56 cm was used. The samples were injected hydrodynamically by 50 mbar applied for 6 s and the separation voltage was 20 kV. The background electrolyte (BGE) was composed with 20 mM sodium borate buffer pH 9 for 120 s. Between the injections, the capillary column was washed for 60 s using BGE.

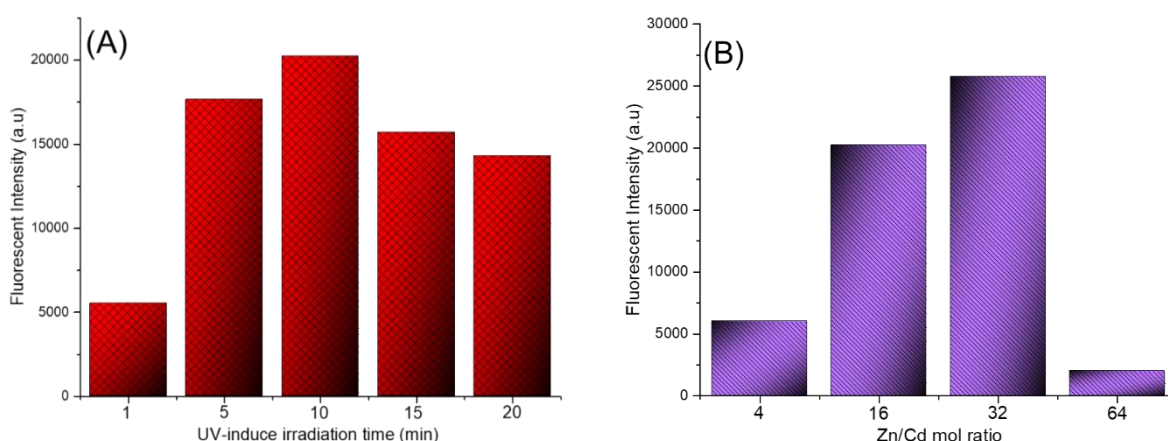
### 2.4. LA-ICP-MS Quantitative Mapping

Qds-Ab formation was evaluated as describer earlier [8] by LA-ICP-MS setup consisting, LA system UP213 (NewWave Research, USA), emitting laser radiation with a wavelength of 213 nm, the pulse width of 4.2 ns, ablation flow of a He (1.0 l/min) into ICP-MS Agilent 7500CE (Agilent Technologies, Japan).

### 3. RESULT AND DISCUSSION

#### 3.1. Effect of UV irradiation time and Zn/Cd molar ratio

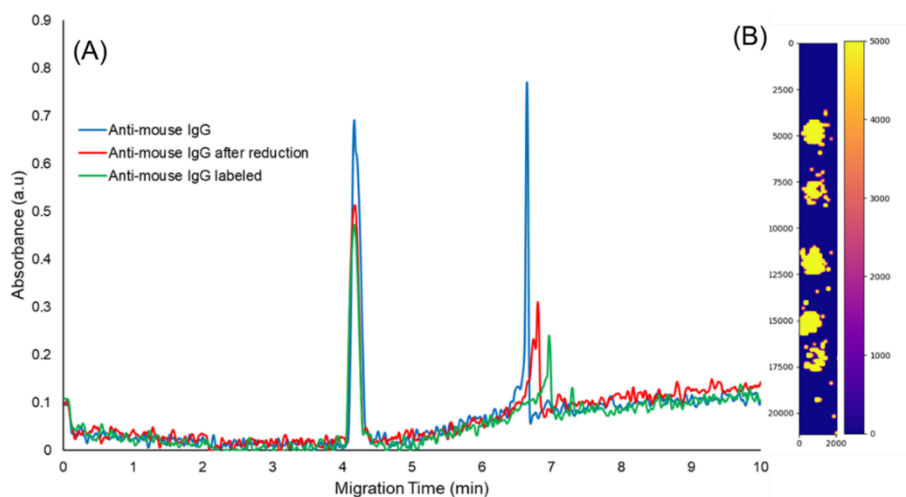
It is comprehended that increasing the UV irradiation time and Zn/Cd molar ratio would increase the yield of prepared QDs-Ab. Therefore, these parameters were evaluated. UV irradiation time had an important effect on the preparation ZnCd/S quantum dots [9] which is illustrated in **Figure 1 (A)**, the maximum fluorescent intensity was achieved at 442 nm under 382 nm excitation. The composition of ZnCd/S quantum dots for obtaining higher fluorescent intensity was mainly dependent on the initial concentration of Zn, Cd, and the intrinsic reactivity of Zn and Cd toward S [10]. The QDs-Ab solution was prepared with the different molar ratios of Zn/Cd (4, 16, 32, and 64) and the maximum emission was obtained in molar ratio 32 as shown in **Figure 1 (B)**.



**Figure 1** Effect of UV-induced irradiation time (A) and Zn/Cd molar ratio on fluorescent intensity (B)

#### 3.2. CE and LA-ICP-MS analysis

To reveal that QDs-Ab was formed, CE is a valuable instrument [11]. **Figure 2 (A)** shows the electropherograms of anti-mouse IgG before and after reduction and anti-mouse IgG labeled by in situ synthesized QDs (QDs-Ab). In general, a peak in migration time 4.17 min belongs to electroosmotic flow (EOF). Firstly, the anti-mouse IgG was analyzed (blue trace) and a sharp peak appeared in the migration time of 6.65 min. Next, anti-mouse IgG after reduction exhibited slightly distorted peak (red trace) with migration time of 6.79 min. Eventually, the peak (green trace) with migration time, 6.96 min demonstrates that anti-mouse IgG was labeled ZnCd/S. The increased migration time suggests not only increasing size but also increasing negative charge of the QDs-Ab.



**Figure 2** Electropherograms of anti-mouse IgG before and after reduction and anti-mouse IgG labeled with ZnCd/S (A) and Cd analysis of QDs-Ab by LA-ICP-MS (B)

LA-ICP-MS has been used for the determination of metals in purified QDs-Ab samples. As is illustrated in **Figure 2 (B)**, five dots of QDs-Ab were detected in polyvinylidene difluoride (PVDF) membrane, with comparable intensity of Cd.

#### 4. CONCLUSION

Labeling anti-mouse IgG with in situ prepared ZnCd/S quantum dots by utilizing antibody thiol groups as a stabilizer was accomplished for the first time. Simplicity, convenience and time-effectiveness are the main benefits of this procedure for antibody labeling. Fluorescent intensity is the main sign of preparation quantum dots optimized based on UV irradiation time and precursor metals ratio. Antibodies labeled with quantum dots were analyzed by LA-ICP-MS and CE and will be used in the near future for immuno-based analysis (e.g. dot blot or ELISA).

#### ACKNOWLEDGEMENTS

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