

PREPARATION OF W-Cu COMPOSITES BY INFILTRATION OF W SKELETONS - REVIEW

Jiří MATĚJÍČEK

Institute of Plasma Physics, Czech Academy of Sciences, Prague, Czech Republic, EU, <u>matejicek@ipp.cas.cz</u>

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Abstract

Tungsten-copper composites feature high corrosion and erosion resistance, very good thermal and electrical conductivity, low thermal expansion and good mechanical properties. They are used in a variety of demanding applications, such as arc-resistant electrodes, high voltage electrical contacts, heat sinks for integrated circuits, etc. They are also foreseen for use in plasma-facing components of fusion reactors, e.g. as a transition layer between the (refractory) plasma-facing tungsten and the (highly conductive) copper-based cooling structure. In general, high density and good bonding of the tungsten and copper phases is desired. Molten copper infiltration into tungsten preforms is among the prospective fabrication technologies; the structure and properties of the resultant composites are dependent on the specific technological parameters.

In this paper, the preparation of W-Cu composites by infiltration of W skeletons is reviewed and attention is paid to the influence of these particular parameters: infiltration temperature, time and atmosphere, tungsten preform porosity, orientation and chemistry (presence or absence of other elements). Optimum parameter combinations for achieving high density and proper bonding of copper and tungsten are identified.

Keywords: Tungsten, copper, composites, molten copper infiltration

1. INTRODUCTION

Tungsten–copper composites combine a number of beneficial properties of both the constituents, such as the excellent thermal and electrical conductivity of copper and the high strength, hardness and erosion resistance and low thermal expansion of tungsten [1]. Thanks to these, they are used in a variety of demanding applications, such as arc-resistant electrodes, high voltage electrical contacts, heat sinks for integrated circuits, high-temperature erosion nozzles, etc. [1,2]. Moreover, W-Cu composites exhibit better machinability in comparison to pure tungsten [1]. These composites are also foreseen for use in plasma-facing components of fusion reactors, where a refractory and erosion-resistant material such as tungsten has to be provided on the plasma facing surface, while a highly conductive material such as copper is needed for the underlying heat sink. The largely different properties and operational temperature windows of these two materials present a major challenge, which can be alleviated by introducing an interlayer with intermediate properties, such as the composite [3]. This interlayer can also have a gradually varying composition, which further reduces the stress concentration [4,5,6].

The W-Cu composites can be produced by a variety of techniques, including cold spraying [7], plasma spraying [8], high-velocity oxy-fuel spraying [9], selective laser melting [10] and various types of powder metallurgy techniques – for example, cold compaction + sintering [11], spark plasma sintering (SPS) [12], pulsed plasma sintering [13], microwave sintering [14], liquid phase sintering [15], hot pressing [16] or powder injection molding (PIM) [17]. A particular technique, making use of the largely different melting points of tungsten and copper, is the molten copper infiltration into pre-sintered tungsten skeletons, which is the focus of this paper.



This technique results in a structure with both phases being contiguous, which can be beneficial for some applications. The contiguity of the phases may also affect the properties of the composite, such as thermal expansion [18] or thermal conductivity [19]. In general, high density of the composite and good bonding of the tungsten and copper phases is desired. This can be to a large extent influenced by the parameters of the fabrication technology. This paper reviews the prospective approaches in liquid copper infiltration into porous tungsten skeletons and focuses on specific parameters which affect the density and interparticle bonding.

2. PREPARATION OF W-CU COMPOSITES BY LIQUID COPPER INFILTRATION

This method generally consists of two steps: preparation of porous W skeleton by sintering and its infiltration by a molten copper. In the following, the role of these specific parameters will be discussed: infiltration temperature, infiltration time, infiltration atmosphere, W skeleton pore size, configuration of the infiltration setup and chemistry (of both the skeleton and the infiltrating liquid). An overview of parameters used in exemplary case studies in provided in **Table 1**.

2.1. Infiltration temperature

The melting point of copper is 1085 °C. However, the experience shows that temperatures just above this value are generally not sufficient for a successful result. For proper filling of the porosity and good bonding of the copper and tungsten phases, proper wetting of the tungsten particles by the liquid copper is necessary. According to [20], the wetting angle is about 25° at 1150 °C, but it gets close to zero in the 1300-1400 °C range. Therefore, this temperature range can be considered optimal for proper wetting when infiltrating with pure copper and it has been used in majority of such experiments listed in **Table 1**. Using lower temperatures typically resulted in lower densities, although the differences were often moderate [20-22]. Still, in [3], infiltration at 1150 °C is reported with apparently fully dense structure. Lower temperatures are generally used with alloys having lower melting point than pure copper, such as Cu-Zn alloy [23] or Cu-based metallic glasses [24, 25], or in combination with additional driving force for the infiltration, such as hot isostatic pressing(HIP) [26] or centrifugal force [3, 27]. Besides the most common heating in a furnace, current heating (similar to spark plasma sintering) can be used; high current densities lead to markedly reduced processing times [30]. Combustion heating with the use of thermite reaction [29-31] can reach temperatures estimated in the 1800 – 2500 °C range.

2.2. Infiltration time

Typical processing times found in the literature (cf. **Table 1**) were in the range of 1-3 h. The required time is necessitated to ensure thorough heating of both constituents, i.e. the skeleton and the infiltrant and to allow its full penetration throughout the structure. The latter factor depends also to some extent on the size of the final product. With the use of rapid heating techniques, the processing times can be significantly reduced – e.g. for the combustion heating, times about 5 min were reported [29], for current-assisted melt infiltration (CAMI) even about 5 s [28]. The latter case resulted in ultra-fast processing, however, the produced composites were inhomogeneous and porous. When the liquid penetration was assisted by the centrifugal force, processing times of the order of 5 min were reported [27].

2.3. Infiltration atmosphere

The atmosphere under which the infiltration is carried out also plays a role. In the literature survey, either vacuum or inert gas (argon, nitrogen) were found, but most often a reducing atmosphere (hydrogen). Reducing atmosphere has the beneficial effect of suppressing the formation of tungsten oxides (or reducing those that may have been already present) [21]. Liquid copper has high contact angles with several oxides, therefore, oxidized W skeleton is wetted less efficiently and this impedes the infiltration process. In [21], sintering and infiltration tests conducted in air were reported to result in no infiltration. Apart from vacuum, atmospheric



pressure was used in practically all works surveyed; only in [26], a higher pressure argon (at 98 MPa) was used in a HIPping step (with the pre-form in an evacuated capsule) to aid the copper penetration. When gases are used, care must be taken to prevent their entrapment in the skeleton, which would hinder the infiltration and result in porosity in the finished product [21]. This can happen during spatially inhomogeneous advancement of the infiltrating copper that may enclose the gases remaining in the skeleton [21,32]. This will be further discussed in the subsequent section.

2.4. Geometric configuration of the infiltration setup

In general, two driving forces can operate during the copper infiltration – gravity and capillarity. Consequently, two variants of the mutual orientation of the copper and tungsten can be used in the most simple arrangements – when the copper is placed on top of the tungsten skeleton, it flows down freely into the pores under the action of gravity; when the skeleton is placed on top, the copper seeps into the pores driven by the capillary forces. In the literature surveyed, both instances were represented equally often (if mentioned at all). In [21], placing the skeleton on top is recommended, to prevent entrapment of the gases if the molten copper flows down faster on the sides of the preform than though the center. Gas entrapment can happen in the opposite configuration as well, as mentioned in [32]. Due to the wall tension, liquid copper may cover the lateral surfaces of the skeleton walls was used in [32]; the composites prepared with this uni-directional infiltration featured higher strength than those prepared without the zirconia layer.

In the more advanced configurations, such as combustion heating + centrifugal infiltration, the tungsten preform is placed at the extremity, in the direction of the centrifugal force, this is followed by the copper and termite (or their mixture) towards the center [29-31]. In [3,33] the centrifugal infiltration concept was used in a cylindrical geometry, when a tungsten preform consisting of braided fibers was placed in a rotating tube, with the copper infiltrating it from the center. This way, composites of tubular shapewere produced. In the CAMI method, Cu powder layer was sandwiched between two W powder layers [28]; because of different porosities, significantly higher current density was estimated for the W powder.

For larger compacts, some spatial variation of the infiltration process and therefore microstructure may occur. For example, in W-Cu compacts of 19 mm height, non-negligible variations in Cu content, W phase contiguity, hardness, elastic modulus and electric conductivity were observed [34].

2.5. Porosity of the W skeleton

To ensure a complete infiltration, the W skeleton needs to have a continuous network of open pores [4]. The size of the pores in the skeleton influences the infiltration process, as the capillary pressure is inversely proportional to the pore diameter [35]. The pore shape plays a role as well – deep and narrow pores take longer to infiltrate. For larger pores, the capillary pressure is smaller, thus they are easier to infiltrate; on the other hand, too large pores may be unable to retail the liquid, for the same reason. In [35], the following pore characteristics are recommended for optimum infiltration: sizes between 1 and 4 μ m, length to radius ratio between 1 and 8.

The pore size is primarily controlled by the size of the W feedstock forming the skeleton (typically powder) and the degree of its sintering. In the surveyed literature (cf. **Table 1**), the most typical W powder sizes were of the order of μ m, but the sizes used ranged from hundreds of nm to tens of μ m, without any obvious limitations for the degree of infiltration. For skeletons prepared from powders of narrow size distribution, the maximum achievable porosity is about 35% (the space among loosely packed powder particles); this can be only reduced by subsequent sintering. If a higher porosity volume – and therefore higher Cu content in the composite – is desired, additional steps need to be taken. In [4], the porosity of the W skeleton was increased by adding an amid wax as a placeholder, which was subsequently removed by debinding at moderate temperature before



the final skeleton sintering. In [36], tungsten oxide was added to W powder to increase the W skeleton porosity; the oxide was reduced to tungsten by sintering in a hydrogen atmosphere. In [37], the porosity was increased by oxidation and volatilization of sintered W skeleton using a mixture of argon and water vapor at elevated temperature. All three approaches permit the formation of porosity gradients. Alternatively, a porosity gradient can be formed by layering of powders of varying size, as the degree of sintering generally increases with decreasing particle size [26, 32, 37, 38]. The porosity can be further controlled by the feedstock geometry. For example, the usage of W preform consisting of braided W fibers resulted in much higher Cu content in the composite than in composites made from W powder [3]. The fiber-reinforced composites also featured higher strength than their powder-based counterparts [39,40].

2.6. Composition

The outcome of the infiltration and thus the properties of the composite can be to a large extent influenced by the composition of both the preform and the infiltrant. Typically, the addition of certain (active) elements to the W feedstock can a) improve the sintering of the skeleton itself, and b) increase the wetting of W particles by the infiltrant, thereby improving the infiltration efficiency [32,41]. These elements are typically Ni, Co, Fe or P and can be applied either as a coating on the W particles or as a powder admixture; in the latter case, their action is facilitated by diffusion during the infiltration process.

Improved wetting by a small Ni additive was demonstrated in [32]. In a similar experiment, Ni was found to be better than Co [42]. In [43], 0.05% addition of Ni was found optimal, while higher Ni content led to higher sintering temperature and lower homogeneity. In [44], W particles coated with Ni, Ni-P and Cu-Ni-P have resulted in composites having higher density, but lower W contiguity. About 10% improvement of density of the infiltrated composite was found in [41] when using Ni or Ni-P coatings on the W powder, leading to reduced electric resistivity. In [39,45], improved contiguity of Cr-coated W fibers and connectivity with neighboring W powder particles was demonstrated. Similar improvement was observed for Ni coating on the W fibers, leading to a marked improvement in tensile strength and electric breakdown strength of the composite [40].

Besides additional elements, the same material as the infiltrant, i.e. copper, can be also added to the W preform. In several studies, the preform was made from a mixture of W and Cu [21,22,45-52]. This approach eases the sintering of the preform and also contributes to increased Cu content, however, can disturb the W contiguity. Application of Cu coating to the W powder has led to superior properties, such as homogeneity, hardness, electric and thermal conductivity and arc erosion resistance, compared to composites made from powder mixtures [48]. Additional reinforcement of the infiltrated W-Cu composite by graphene results in improved uniformity and arc erosion resistance [53], while a Cu-coated graphene prevents interfacial reactions between tungsten and graphene and improves the structural homogeneity [54].

As mentioned before, replacement of copper with alloys having lower melting point permits using lower infiltration temperatures. For example, in [23], Cu80-Zn20 alloy was infiltrated at 1100 °C into a W fiber preform with adequate density. For Cu-based metallic glasses, infiltration temperatures in the 850-900 °C range were found optimal forCu₄₇Ti₃₃Zr₁₁Ni₆Sn₂Si₁ [24] and around 965 °C for (Cu₅₀Zr₄₃Al₇)_{99.5}Si_{0.5}[25].

3. CONCLUSION

In this article, the preparation of W-Cu composites by liquid Cu infiltration of W skeletons was reviewed. The role if specific technological parameters, namely the infiltration temperature, time and atmosphere, geometric arrangement, skeleton porosity and composition of the preform and infiltrant, was addressed. From a number of surveyed case studies, the suitable parameter combinations for achieving high density and good bonding of the W and Cu phases were pinpointed. These results can aid in the preparation of such composites for demanding applications or in further developments of the technique.



Table 1Overview of technological parameters used in the surveyed case studies on Cu-infiltration of Wskeletons. Ref. = reference, Preform = characteristics of the preform, Cu infiltration – parameters ofthe infiltration, Atm. = atmosphere used during the infiltration, Main advantage – factor(s) of aparticular setup that are considered advantageous

Ref.	Preform	Cu infiltration	Atm.	Main advantage
[32]	W 6 μm, sintered at 1400-2150 °C, H2, ~78% density W+0.05%Ni milled +ZrO2 wall to reduce wall tension	1300 °C, 2h	H ₂	 Ni additive - improved wetting ZrO₂ wall -> directional infiltration, no trapped gas
[42]	W 6 μm + 0.05, 0.25, 0.5% Ni or Co	1250-1300 °C, 2h	H ₂	additives for improved wetting
[44]	W 8 μm, electroless-plated by Ni, Ni-P, Ni-Cu-P	1300 °C	H ₂	coating for better wetting
[28]	W 100 nm, Cu 45 μm	current-activated melt infiltration	Ar	rapid heating
[35]	W ~10 μm, sintered at 1330 °C/5h + 1790 °C/5h, H2	1400 °C, graphite boat, 2h	H ₂	
[20]	W fiber preforms	1200, 1300 °C, 1h, alumina crucible		
[43]	W <6 μm + 0.05, 0.25, 0.5% Ni, sintered at 1250-1550 °C, 4h, H2	1300 °C, 1h	H2	Ni additive for improved wetting
[55]	W <6 μm, sintered at 1550 °C, 4h, H2, p=200-663 MPa	1300 °C, 1h	H2	higher p -> lower T for W sintering
[30, 31]	W fibers + Cu powder preform + thermite	Combustion synthesis melt infiltration under ultra-high gravity, ~2550 °C	vacuum 100 Pa	centrifugal force + high T
[46]	W 1.3 μm + Cu 1 μm; sintered at 1000 °C, 2h	1300 °C, 3h	H ₂	
[21]	W 1 μm, Cu 10 μm PIM preforms	1150, 1200, 1250 °C, 1h	H ₂	
[23]	W fibers	1100 °C, 4h (Cu-Zn alloy)	H ₂	
[47]	W 5-7 μm, Cu <80 μm; 80:20 and 70:30	1350 °C, 90min	H ₂	cold pressing of powder mixture + infiltration sintering
[48]	W 6-8 μm + Cu mixture; Cu- coated W 12-14 μm	1350 °C, 90min	H ₂	cold pressing of powder + infiltration sintering
[26]	graded W (1-3 μm) skeleton, sintered at 1770 °C, 0.1 MPa, Ar, 8h, open HIPped at 1800 °C, 181 MPa, Ar, 4h	1100 °C, 2h	Ar	evacuated HIP can, Ar pressure 98MPa
[24]	W wires 800 µm	800-1000 °C, 10-30min (Cu- based metallic glass)	Ar	
[39]	W wires, coated by Cr by PVD + W+Cu+Ni powder; pre-sintered	1300 °C, 2h	H ₂	Cr coating
[22]	W 4 μm + 24% Cu <63 μm + 1% Ni 5 μm, pressed and sintered at 950 °C, 30 min	1150 °C, 30min	H ₂ , N ₂	



[37]	W skeletons with porosity gradient prepared by oxidation and volatilization	no details		porosity gradient
[41]	W 8 μm, electroless-plated by Ni, Ni-P, pressed at 348 MPa	1300 °C, 1 h	N2/H2 80/20	coating for better wetting
[3]	W skeleton, cold pressed + sintered at 1150 $^\circ\text{C},$ 2 h, H_2	1150 °C, 2 h	H ₂	
[3, 33]	braided W fibers, 50 mm	centrifugal melt infiltration		centrifugal melt infiltration for tubular shape composite
[25]	W wires	965, 990, 1015 °C, 10, 15 min (Cu-based metallic glass)		
[36]	W powder 2 μ m + 5%, 10%, 15% BTO powder 11 μ m, cold pressed + sintered at 1400 °C, H ₂	1300 °C, 1 h	H ₂	blue tungsten oxide (BTO) addition to increase W skeleton porosity
[49]	W 10 μm + 5%, 8%, 20% Cu 22 μm mixture, pressed	microwave vacuum infiltration sintering, 1200 °C, 3 h	vacuum	
[4]	W 4 μm + amid wax, pressed + sintered at 1200 °C, 1 h, H ₂	1200 °C, 30 min (Cu1CrZr alloy); Cu1CrZr plate on top		additive for porosity control
[50]	W 8 μm, 800, 600, 400 nm + Cu 50 μm, 75:25, pressed at 340 MPa	sintering+infiltration 1350 °C, 40 min	H ₂	fine W powders -> bi- continuous structure
[27]	W 1-6 μm, pressed, sintered at 1350 °C, 1.5h, H ₂	centrifugal infiltration, 1150 °C, 5 min	vacuum or added flux	
[51]	W 500 nm + Cu 50-70 μm, 75:25, pressed + sintered at 1050, 1150, 1250, 1350 °C, 2 h, H ₂	2 h	H2	
[29]	W 3 μm + Cu 3 μm blends + thermite	high-gravity combustion synthesis and melt- infiltration, T~1800 °C, 5 min		centrifugal force + high T
[54]	W 5-7 μm + Cu 3-5 μm + Cu- coated graphene, pressed	Cu block on top, 1300 °C, 90 min	Ar	
[40]	W powder 4-6 μm + Cu powder 50-70 μm + W fibers (uncoated or Ni-coated by electrolessplating), pressed + sintered at 1300 °C, 2 h	1300 °C, 2 h	H ₂	Ni coating for reduced porosity
[45]	W powder 500 nm + Cu powder 50-70 μm + W fibers (Cr-coated by magnetron sputtering), pressed + sintered at 1350 °C, 2 h	1350 °C, 2 h	H ₂	Cr coating for improved sintering
[52]	W powder 4-6 μm + Cu powder 50-70 μm + W fibers (Ti-coated by magnetron sputtering), pressed + sintered at 1350 °C, 2h	1300 °C, 2 h (Cu+0.5%Cr)	H2	Ti coating for improved sintering
[34]	W powder, pressed at 300 MPa, sintered at 1550 °C, 4h, H ₂	1300 °C, 1h, Cu on top	H ₂	study of spatial homogeneity



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