

## THE EFFECT OF Al AND C CONTENTS VARIATIONS ON THE PHASE COMPOSITION AND PHASE MORPHOLOGY OF PYROFERAL<sup>®</sup> TYPE IRON ALUMINIDES

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### Abstract

Complex study of alloys based on FeAluminide was leading in the fifties of the last century to the proposal of. The composition of the binary Fe-Al alloy was modified by C and Si. This alloy with excellent corrosion and heat resistant properties was used to replace the high chromium and nickel alloyed cast iron.

The investigations performed recently on Fe-40 at.% Al type alloys with carbon contents  $1.9 \div 3.8$  at.% and aluminum up to 43 at.% were designed to explain in more detail the origin of the structure and mechanical properties of the mentioned material.

It is the purpose of this contribution describes the effect of Al and C on the structure of iron aluminides with the composition of Pyroferal<sup>®</sup>. The appearing and morphology of carbide  $Al_4C_3$  and of graphite is described in the as cast and annealed state.

**Keywords:** Intermetallic alloys, iron aluminides, Pyroferal

### 1. INTRODUCTION

A possible method for improvement of creep resistant FeAl-type aluminides is the strengthening by second phase particles. One possibility is the addition of carbon. The strengthening is due to the formation of carbide particles, for example  $Fe_3AlC_n$  perovskite-type carbide or  $Al_4C_3$  carbide.

On the same basis the alloy Pyroferal<sup>®</sup> ( $44.6 \div 46.5$  at.% Al and  $3.4 \div 4.0$  at.% C) was prepared in Czech republic as replacement of expensive nickel and chromium steels in the fifties of last century, analogous to Thermagal<sup>®</sup> in France or Tchugal<sup>®</sup> in Soviet Union. There are available very important results, which were achieved during the research between 1950 and 1962, based on the numerous research reports of Prof. J. Pluhař, Ing. M. Vyklický and Ing. H. Tůma, the scientists, who initiated and completed the research leading to the development of Pyroferal<sup>®</sup> (members of the VÚMT Prague - the predecessor of the present SVÚM Běchovice a.s.).

Pyroferal<sup>®</sup> offered quite impressive results in high temperature corrosion resistance. It was tested against air atmosphere, vanadium pentoxide, molten glass, carburization, nitration and the atmosphere of the natural gas cracking generators. Good results were also described in wear resistance and hardness due to appearance of hard and brittle aluminum carbide  $Al_4C_3$  [1, 2].

The recent attention was given to FeAl based alloys with composition similar to Pyroferal<sup>®</sup>. The effect of carbon addition on the phase composition and morphology [3, 4] and the effect on the high-temperature deformation properties [5] were discussed.

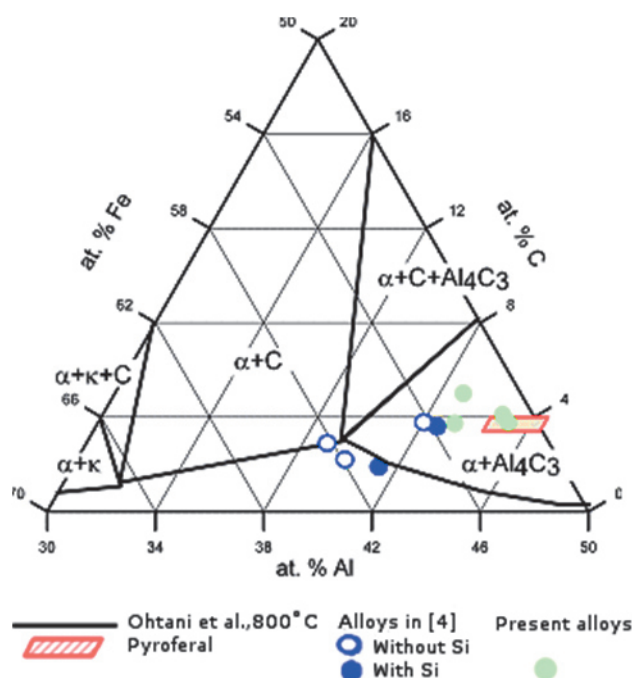
The present paper completes structural studies of alloys with composition within range of Al and C close and identical as Pyroferal<sup>®</sup>. The main task was to find the effect of deviations of Al and C content compared to Pyroferal<sup>®</sup>.

## 2. MATERIALS AND EXPERIMENTAL METHODS

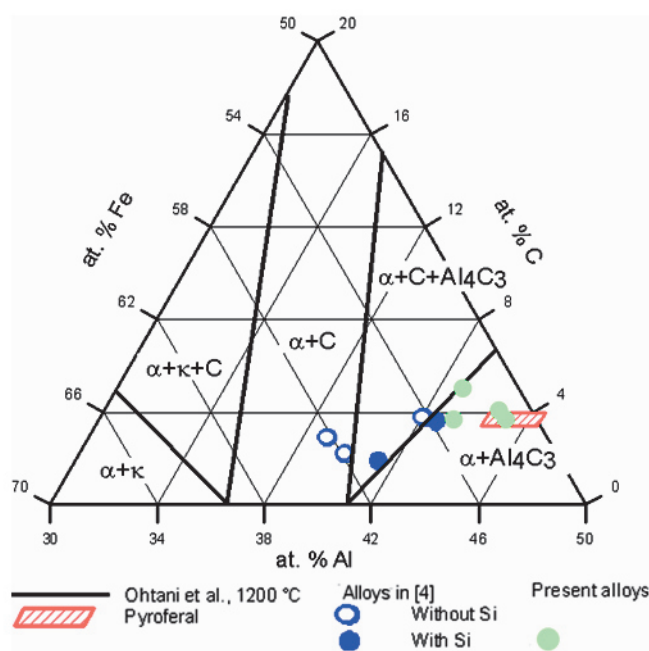
The research was performed with alloys of the composition given in **Table 1**. Compositions are also given in ternary diagram in **Fig. 1 a, b**.

**Table 1** The chemical composition of the alloys (at.%)

Alloy	Fe	Al	C	Si	Denoted
FA 3.7C Si	balance	45.1	3.7	0.8	P
FA 3.8C Si	balance	43.3	3.8	0.8	N
FA 4.0C Si	balance	44.7	4.0	0.8	R
FA 4.9C Si	balance	43.0	4.9	0.7	O



**Fig. 1a** Ternary Fe-Al-C phase diagram [6] for 800 °C with concentrations within this paper and corresponding to [4]



**Fig. 1b** Isothermal section of ternary phase diagram via Ohtani for 1200 °C with composition point

All alloys were prepared by vacuum melting in the induction furnace. The cast slabs (cross section 19.5x 40 mm, length 200 mm) were rolled at 1230 °C in the five steps on the laboratory rolling-mill K350.

The chemical composition of all alloys was determined using spectral analysis by optical emission spectrometer LECO GDS 750a.

The morphology of phases was studied by light optical microscopy (LOM) with Nomarski-contrast application and scanning electron microscopy (SEM). The phases were determined by energy dispersive X-ray spectroscopy and X-ray diffraction analysis (XRD).

The heat treatment with different temperature, time and cooling was used:

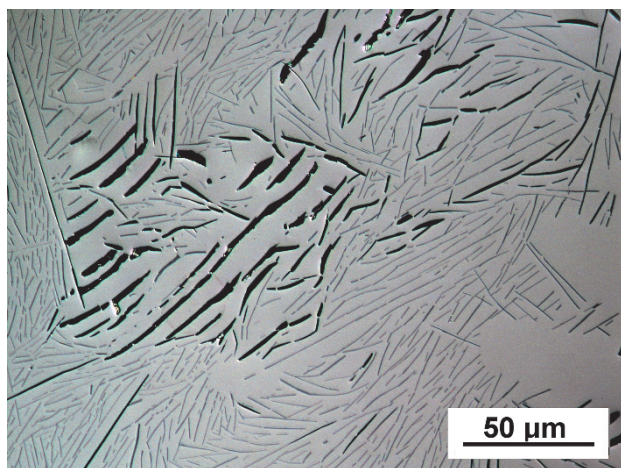
- annealing at 800 °C 8 h, quenching into the mineral oil,
- annealing at 1100 °C 24 h, quenching into the mineral oil,
- long-time annealing at 1150 °C 200 h, cooling in air.

### 3. RESULTS AND DISCUSSION

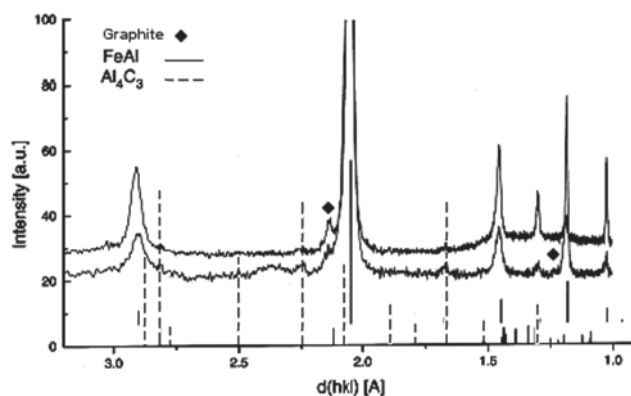
The composition of alloys P and R is very similar with respect to carbon and aluminum contents; the content of silicon is almost identical. To see the effect of Al content on the morphology and structure of phases these two alloys can be compared with alloy N containing less Al. To observe the effect of C content on phase structure, the alloy N is compared with alloy O.

#### 3.1. The effect of Al content on the phase structure (the comparing of alloys P and R to N)

The increasing Al content has no effect on the phase composition of alloys in as cast state - it is identical in all alloys. Three phases (**Fig. 2**, XRD in **Fig. 3**) are present - eutectic mixture of the solid solution  $\alpha$  (bright area in **Fig. 2**) and fine needle-like  $\text{Al}_4\text{C}_3$  phase, and clumps of coarse graphite lamellae (size 50-200  $\mu\text{m}$ ) - see detail in **Fig. 2**.



**Fig. 2** Alloy P, as cast state (bright matrix, fine lamellae of  $\text{Al}_4\text{C}_3$  carbide and black graphite lamellae), LOM



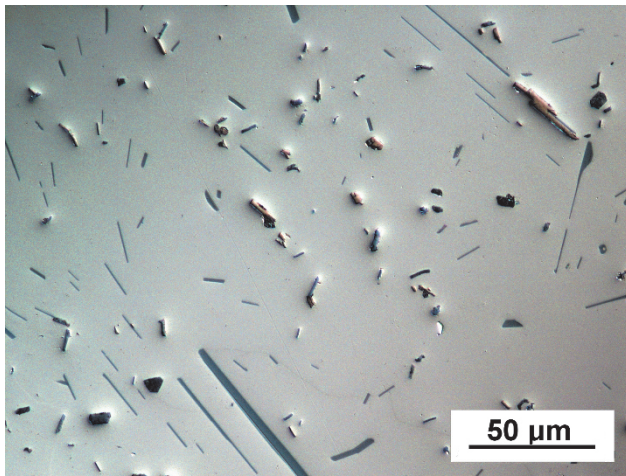
**Fig. 3** The diffraction curve of as cast (lower curve) and annealed state (upper) of alloy N

The differences appear in the structure after high-temperature heat treatment (annealing at 1100 °C and 1150 °C): graphite lamellae are totally dissolved in alloys R and P, whereas graphite flakes remain in the structure of alloy N - compare **Figs. 4** and **5**. EDX and XRD analyse were used for verification of graphite and carbide phase. A small volume fraction of needle-like carbide phase is conserved in the structures of all alloys after annealing 1100 °C /24 h/ oil, long time annealing leads to gradual solution needle-like carbide phase - see **Figs. 4, 5**. It is obvious that bigger carbide needles are getting coarsen during annealing - **Fig. 5**.

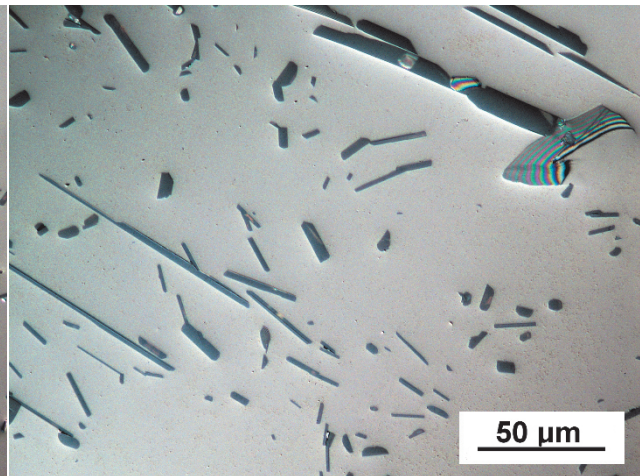
Higher amount of coarse carbide lamellae are present in the structures of alloys R and P (higher Al content) than in the structure of alloy N (lower Al content) - compare **Figs. 4** and **5**.

Higher Al content (more than 44 at.%) for Pyroferal© type alloys may be recommended on the basis of the comparison of structure: the higher Al content leads to conservation of carbide  $\text{Al}_4\text{C}_3$  after annealing, whereas lower Al content leads to conservation of the graphite phase. The structure with higher quantity fine needle-like  $\text{Al}_4\text{C}_3$  phase is advantageous for high - temperature properties of iron aluminides [5] as well as absence of the soft graphite phase.





**Fig. 4** Alloy N, after annealing 1150 °C /200 h/ air,  $\text{Al}_4\text{C}_3$  lamellae, graphite in flakes, LOM

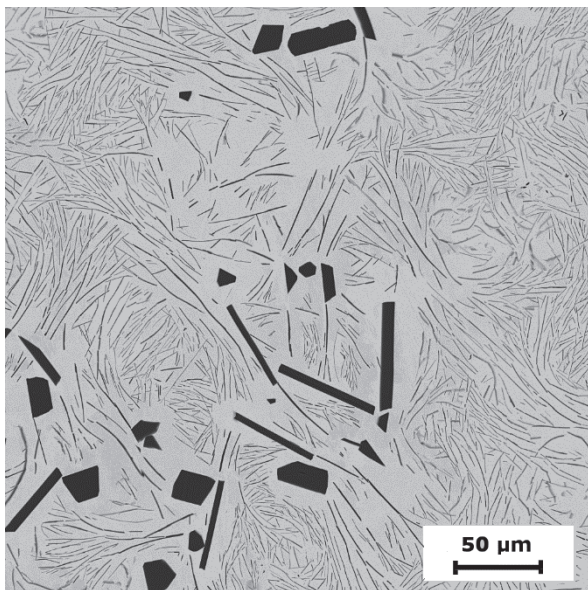


**Fig. 5** Alloy R, after annealing 1150 °C /200h/ air,  $\text{Al}_4\text{C}_3$  lamellae, LOM

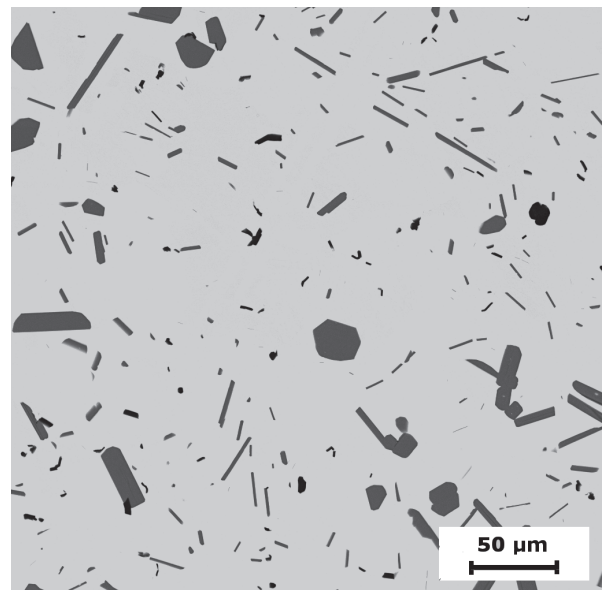
### 3.2. The effect of C content on the phase structure (the comparing of alloys N and O)

The alloy N and O have comparable Al content, but marked difference of carbon content - alloy O has higher C content than alloy N. Phase composition of alloy O is the same as of N alloy: solid solution FeAl,  $\text{Al}_4\text{C}_3$  carbide and graphite (verified by XRD) - but morphology of phases is very different. The carbide  $\text{Al}_4\text{C}_3$  occurs only as fine needle-like phase in the as cast structure of alloy N with lower C, whereas coarse carbide plates and fine needle-like carbide are present in the structure of alloy O with high C (**Fig. 6**). Graphite is present as lamellae and flakes in both alloys.

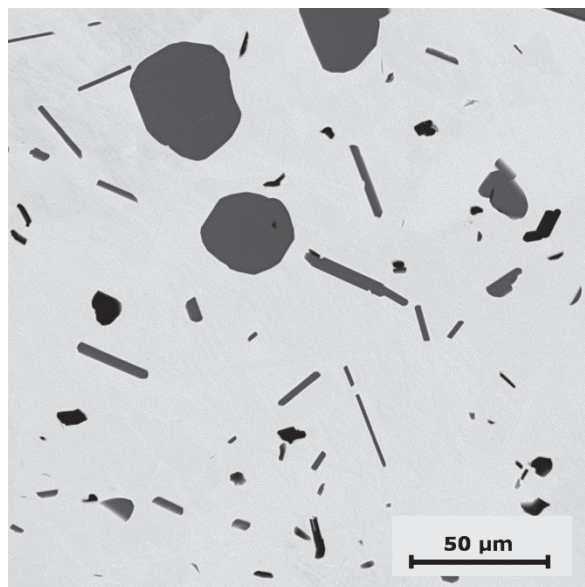
Differences are much more pronounced after heat treatment: the fine needle-like  $\text{Al}_4\text{C}_3$  dissolves gradually, but flakes of graphite remain in alloy N (lower C). In the structure of alloy O (high C) fine needle-like phase dissolves gradually too, but coarse carbide plates are coarsening even more (compare **Figs. 6 and 7**). Small amount of graphite flakes remain in the structure of alloy O too.



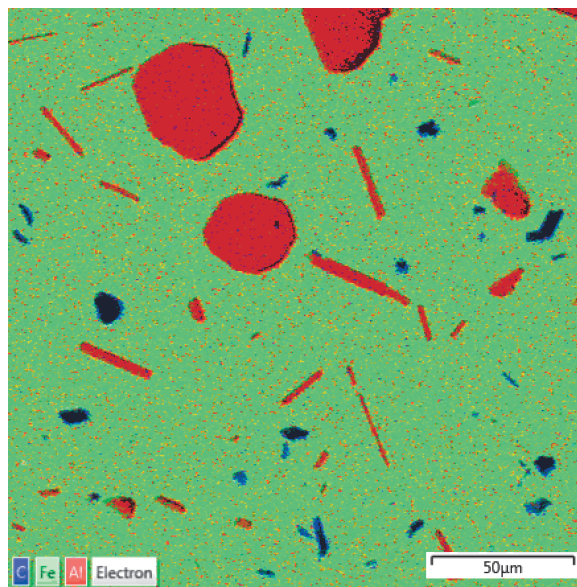
**Fig. 6** Alloy O, as received,  $\text{Al}_4\text{C}_3$  as plates and fine needle-like phase, BSE image



**Fig. 7** Alloy O, after annealing 1150 °C/200 h,  $\text{Al}_4\text{C}_3$  as plates and coarse needle, graphite, BSE image



**Fig. 8a** Alloy O, after annealing 1150 °C/200 h, BSE image - small black particles are graphite, dark grey  $\text{Al}_4\text{C}_3$



**Fig. 8b** Alloy O, after annealing 1150 °C/200 h, EDX map image - phase verification: small blue particles are graphite flakes, red particles are  $\text{Al}_4\text{C}_3$

The morphology of graphite particles can be similar to the morphology of  $\text{Al}_4\text{C}_3$  carbide (lamellae or oval-shaped particles of different sizes) after high-temperature annealing, and differentiation of both phases is very difficult for this reason. Energy Dispersive X-ray analysis was used for phase verification - see **Fig. 8a, 8b**.

The high C content leads to the formation of carbide plates in as cast state and to further coarsening during heat treatment. These brittle carbide plates introduce into the matrix internal stresses and decrease brittle fracture resistivity. Their high hardness increases wear resistance of alloys.

### 3.3. The phases in the investigated alloys

The main problem of phase composition is presence  $\text{Al}_4\text{C}_3$  carbide. It is given in older version of Fe-Al-C diagram by Vogel and Mader [7] - they indicated this carbide at concentration about 44 at.% Al and 1 at.% C. It is in agreement with older experiments of Vyklický and Tůma [1]. This phase is not mentioned for such concentrations later by Nishida [8] and recently by Palm and Inden [9]. The addition of silicon is supposed to support the formation of  $\text{Al}_4\text{C}_3$  carbide [4].

The summary of the phase composition of the alloys after heat treatment is given in **Table 2**. The presence of phases is compared to ternary diagram of Ohtani et al. [6] (800 °C and 1200 °C), see **Figs. 1a** and **1b**.

**Table 2** The summary of the phase composition of the alloys after heat treatment

Alloy	Ohtani 800 °C	Present paper 800 °C	Ohtani 1200 °C	Present paper 1150 °C
N	FeAl + $\text{Al}_4\text{C}_3$	FeAl+ $\text{Al}_4\text{C}_3$ +(C <sub>F</sub> )	FeAl + $\text{Al}_4\text{C}_3$	FeAl+ $\text{Al}_4\text{C}_3$ +(C <sub>F</sub> )
P	FeAl + $\text{Al}_4\text{C}_3$	FeAl+ $\text{Al}_4\text{C}_3$ +(C <sub>L</sub> )	FeAl + $\text{Al}_4\text{C}_3$	FeAl + $\text{Al}_4\text{C}_3$
R	FeAl + $\text{Al}_4\text{C}_3$	FeAl+ $\text{Al}_4\text{C}_3$ +(C <sub>L</sub> )	FeAl + $\text{Al}_4\text{C}_3$	FeAl + $\text{Al}_4\text{C}_3$
O	FeAl + $\text{Al}_4\text{C}_3$	FeAl+ $\text{Al}_4\text{C}_3$ +(C <sub>F</sub> )	FeAl+ $\text{Al}_4\text{C}_3$ +(C)	FeAl+ $\text{Al}_4\text{C}_3$ +(C <sub>F</sub> )

(C) ...minority phase, C<sub>F</sub>....graphite in flakes form, C<sub>L</sub>....graphite in lamellae form

It is obvious, that graphite is present in the structure an extra phase if compared to Ohtani diagrams. Its presence in the structure is reported by Palm and Inden [9].



The annealing at 800 °C/8 h induced refinement of structure for all alloys, i.e. the solution of coarse graphite lamellae compared to as cast. The residual graphite in the structure of all alloy as lamellae (alloys P and R) or flakes (alloys N and O) was observed - see **Table 2**.

The long-time annealing at 1150 °C /200 h induces redistribution of  $\text{Al}_4\text{C}_3$  needles (alloy N). Very pronounced coarsening of  $\text{Al}_4\text{C}_3$  plates is observed in alloy O. Little graphite flakes were conserved in both alloys with lower Al content (N, O). The graphite was totally dissolved in alloys with higher Al content (P, R). The coarsening of needle-like phase  $\text{Al}_4\text{C}_3$  was noticed in these alloys (**Fig. 5**).

## CONCLUSIONS

- reduction of Al content below 44 at.% leads to the presence of the soft graphite phase in the structure after high-temperature (both 800 and 1150 °C) heat treatment,
- increase of C content over 4 at.% leads to the formation of very coarse and brittle phase -  $\text{Al}_4\text{C}_3$  plates, which is disadvantageous for deformation properties,
- annealing at 800 °C contributes to dissolution or to refinement of coarse graphite phase in alloys P,R and N,
- the presence of  $\text{Al}_4\text{C}_3$  carbide corresponds to the ternary diagram of Ohtani [6],
- the presence of the residual graphite is in agreement with the ternary diagram of Palm and Inden [9].

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## REFERENCES

- [1] VYKLIČKÝ M., TŮMA H. Krystalisace technických slitin Fe-Al-C v oblasti tuhého roztoku  $\alpha$ , Hutnické listy, Vol. 14, 1959, p. 118.
- [2] PLUHAŘ J., VYKLIČKÝ M. Vlastnosti žáruvzdorné slitiny ČSN 42 2484 (Pyroferal) a její použití (Properties of heat-resistant alloy ČSN 42 2484 (Pyroferal) and its application). Technical report No. 189. Ministry of Heavy Industry, Prague, 1959.
- [3] VODIČKOVÁ V., KRATOCHVÍL P., DOBEŠ F. The effect of carbon on the structure of Fe-40Al-xC (at.%) ternary alloys. Kovove Mater. Vol. 45, 2007, pp. 153-158.
- [4] KRATOCHVÍL P., DOBEŠ F., VODIČKOVÁ V. The effect of silicon on the structure of Fe-40 at.% Al type alloys with high contents of carbon. Intermetallics, Vol. 17, 2009, pp. 39-45.
- [5] DOBEŠ F., KRATOCHVÍL P., VODIČKOVÁ V., MILIČKA K., PEŠIČKA J. The effect of carbon on high temperature deformation of Fe-40Al-xC (at.%) ternary alloys. Kovove Mater., Vol. 46, No. 2, 2008, pp. 97-103.
- [6] OHTANI H., YAMANO M., HASEBE M. Thermodynamic analysis of the Fe-Al-C ternary system by incorporating ab initio energetic calculations into the CALPHAD approach. ISIJ Int., Vol. 44, 2004, p. 1738.
- [7] VOGEL R., MÄDER H., Arch. Eisenhüttenw., Vol. 9, 1936, p. 333-340.
- [8] NISHIDA K. Technical reports of the Hokaido University, Vol. 48, 1968, p. 79.
- [9] PALM M., INDEN G. Experimental determination of phase equilibria in the Fe-Al-C system. Intermetallics, Vol. 3, 1995, pp. 443-454.