

THE KINETICS OF PHASE TRANSFORMATIONS DURING CONTINUOUS HEATING FROM AS-QUENCHED STATE OF 17-4PH STEEL

BAŁA Piotr^{1,2}, KRAWCZYK Janusz¹, PAWŁOWSKI Bogdan¹, DZIURKA Rafał¹, CIOS Grzegorz²

¹AGH University of Science and Technology, Faculty of Metals Engineering and Industrial Computer Science, Cracow, Poland, EU

²AGH University of Science and Technology, Academic Centre of Materials and Nanotechnology, Cracow, Poland, EU

Abstract

This work contains a detailed description of the influence of continuous heating from as-quenched state on microstructure development in the 17-4PH steel. The temperature ranges of the phase transformations occurring during heating from the as-quenched state (tempering) were determined from dilatometric tests. An interpretation of heating dilatometric curves was also presented. The temperature ranges of the occurred transformations during tempering of investigated steel were drawn on CHT (Continuous Heating Transformations) diagram. Moreover, the microstructure development in investigated samples, reflecting the extension of the phase transformations during tempering, was discussed as well. In as-quenched state the microstructure of investigated steel consist of low carbon martensite, ferrite δ banding, and niobium secondary carbides of MC type, which have not been dissolved during austenitizing. Changes in the microstructure of the investigated steel, during the heating from the as-quenched state result in precipitation of transition iron carbides, transformation of small amount of retained austenite into martensite, and precipitation of the Cu_e phase. There were no significant changes in the chemical composition as well as in the morphology of the δ -ferrite.

Keywords: 17-4PH steel, tempering, kinetics, phase transformations, CHT diagram

1. INTRODUCTION

The study of phase transformations in alloys, a specially steels, is essential of primary importance for the optimization of thermal and thermomechanical treatments. One of the most popular techniques is dilatometry. A simple principle which consists of heating and/or cooling a material while measuring dimensional changes as a function of temperature or time has been studied. Dilatometry allows monitoring of any bulk phase transformation occurring in metals and alloys, such transformations being detected as soon as the densities of phases differ [1-8]. Dilatometry is commonly used for intermetallic phase precipitation kinetics analysis of precipitation hardening alloys, such as nickel based alloys [8-11].

Precipitation-hardened stainless steels were first developed during the 1940s, and since then, they have become increasingly important in a variety of applications in which their special properties can be utilized. The most important of these properties are ease of fabrication, high strength, relatively good ductility, and excellent corrosion resistance. The 17-4 PH (precipitation hardening) stainless steel is a martensitic stainless steel containing up to 5 wt. % Cu and is strengthened by the precipitation of highly dispersed copper particles in the low carbon martensite matrix. This steel is more common than any other type of precipitation hardened stainless steels and thus has been used for variety of applications including oil field valve parts, chemical process equipment, aircraft fittings, pump shafts, gears, paper mill equipment, and jet engine parts [12-15].

2. EXPERIMENTAL

The research was performed on 17-4PH steel. The chemical composition of investigated steel is given in **Table 1**. The dilatometric investigations were performed using the Adamel DT1000 high-speed dilatometer. Specimens of 2 mm in diameter and height of 12 mm, were austenitized for 20 minutes at 1040 °C, quenched and heated to 700°C at a rate ranging between 0.05 and 35 °C/s. Additionally, the quenched samples were heated with a rate of 0.05 °C/s up to chosen temperatures. The microstructure of the investigated steel was examined by a light microscope Axiovert 200 MAT and the Zeiss scanning electron microscope with Tident XM 4 system of the EDAX Company - energy dispersive X-RAY (EDS) analysis and JEM 200CX transmission electron microscope.

Table 1 Chemical composition of the investigated steel (wt. %)

Grade	C	Mn	Si	P	S	Cr	Ni	Nb	Cu	Fe
17-4PH	0.04	0.88	0.32	0.022	0.002	15.96	4.31	0.26	3.53	bal.

The microstructure of the investigated steel in as-delivered condition is presented in **Fig. 1**. The microstructure consist of tempered martensite, ferrite δ banding (**Fig. 1b**) as well as niobium secondary carbides of MC type. Hardness of the steel in as-delivered condition is 366 HV10. Microstructure of investigated steel after quenching from 1040 °C is presented in **Fig. 2**. In as-quenched state the microstructure of the investigated steel consist of low carbon martensite, ferrite δ banding (**Fig. 2b**) as well as niobium secondary carbides of MC type which have not been dissolved during austenitizing (**Fig. 2c**). Hardness after quenching is close to the hardness in as-delivered condition, and equal 360 HV10.

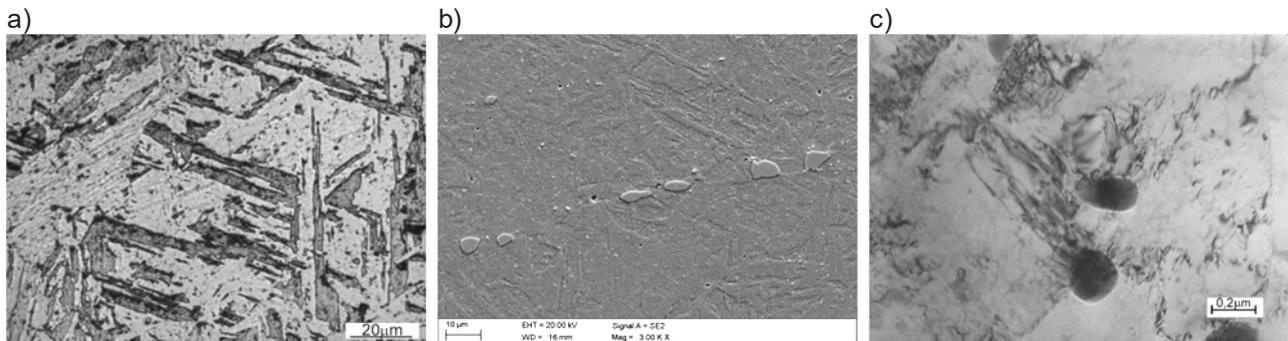


Fig. 1 Microstructures of the 17-4PH steel in as-delivered condition: a) Light microscope, b) SEM, c) TEM

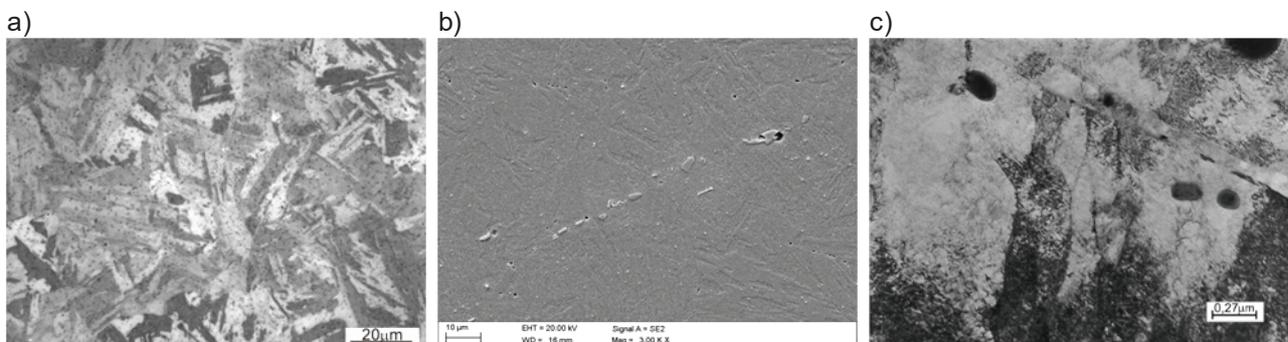


Fig. 2 Microstructures of the 17-4PH steel after quenching from 1040 °C: a) Light microscope, b) SEM, c) TEM

3. RESULTS AND DISCUSSION

Fig. 3 shows a dilatogram of the quenched samples heated with a rate of 0.5 °C/s, along with the corresponding differential curve showing a method of an interpretation of dilatograms, basing on which a CHT diagrams were created. It is apparent that during, the first stage of tempering, the investigated steel exhibits shrinkage, which may mainly be attributed to transition iron carbide precipitation. This shrinkage starts at ϵ_s temperature, and ends at ϵ_f one. When the ϵ_f temperature is reached, a positive dilatation effect connected with retained austenite transformation can also be noticed. This effect is noticeable in the temperature range of retained austenite starts (RA_s) + retained austenite finish (RA_f). Both above mentioned effects, especially dilatation effect related to retained austenite transformation, are not intensive. Between $(Cu_\epsilon)_s$ and $(Cu_\epsilon)_f$ temperatures the precipitation of copper rich phase (Cu_ϵ) takes place. **Fig. 4** shows CHT diagram for the investigated steel. The diagram shows the ranges of iron transition carbide and the temperature range for the transformation of the retained austenite as well as the temperature range of precipitation of copper rich phase (Cu_ϵ). As can be noticed, transformation start and transformation finish temperatures increase with an increasing heating rate from 0.05 to 35 °C/s.

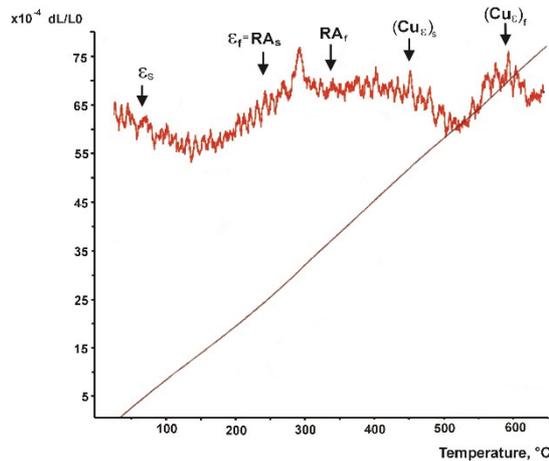


Fig. 3 The dilatometric curve of heating with the rate of 0.5 °C/s as well as the corresponding differential curve with marked interpretation of phase transformation occurred in the 17-4PH steel during heating from as-quenched state

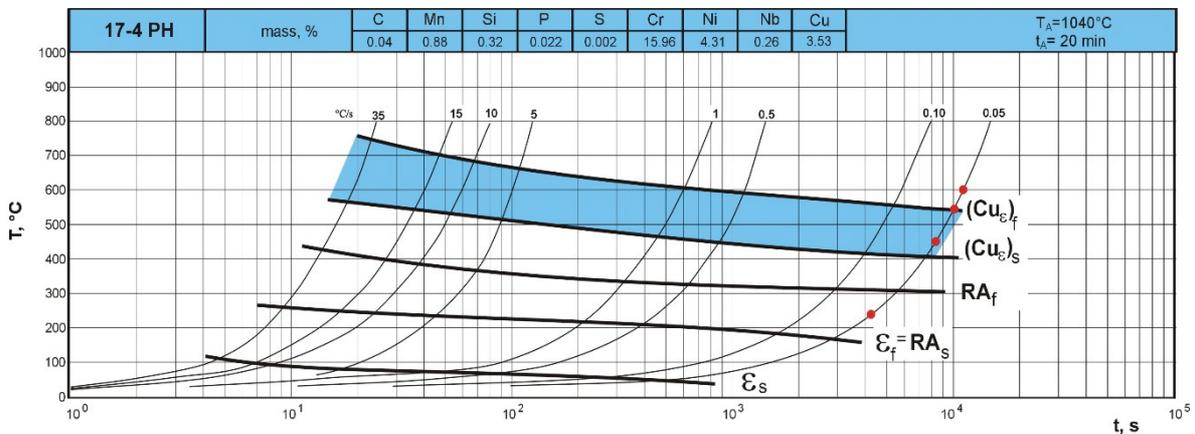


Fig. 4 CHT diagram for investigated steel. The temperatures of the end of heating the samples for metallographic investigation are marked red

Continuous heating from as-quenched state up to 240 °C and 450 °C (**Fig. 5**) does not caused visible changes in the microstructure of investigated steel. But it is worth to note, that heating up to 450 °C increased hardness to 396 HV10. It is results of precipitation starts of copper rich phase (Cu_ϵ) - see **Fig. 4**. Unfortunately we were not able to find it in the microstructure, even by use TEM. Visible changes in the microstructure of the investigated steel were caused by heating it up to 540 °C (**Fig. 6**) and 600 °C (**Fig. 7**). There are clearly visible grain boundaries of a prior austenite and the morphology characteristic for tempered martensite. SEM image presented in **Fig. 6b** and TEM images included in **Figs. 6c and 7c** show precipitates of copper rich phase (Cu_ϵ).

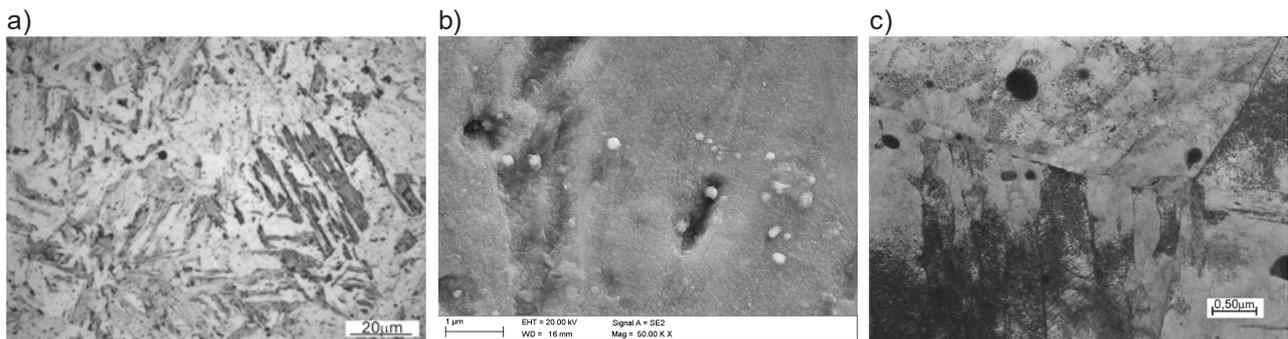


Fig. 5 The microstructure of investigated 17-4PH steel after quenching from 1040 °C and heated up to 450 °C with the rate of 0.5 °C/s: a) light microscope, b) SEM; c) TEM

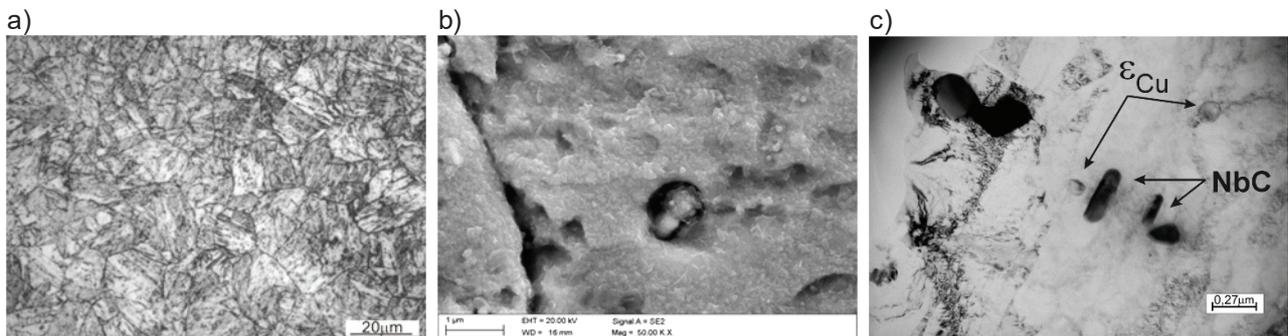


Fig. 6 The microstructure of investigated 17-4PH steel after quenching from 1040 °C and heated up to 540 °C with the rate of 0.5 °C/s: a) light microscope, b) SEM; c) TEM

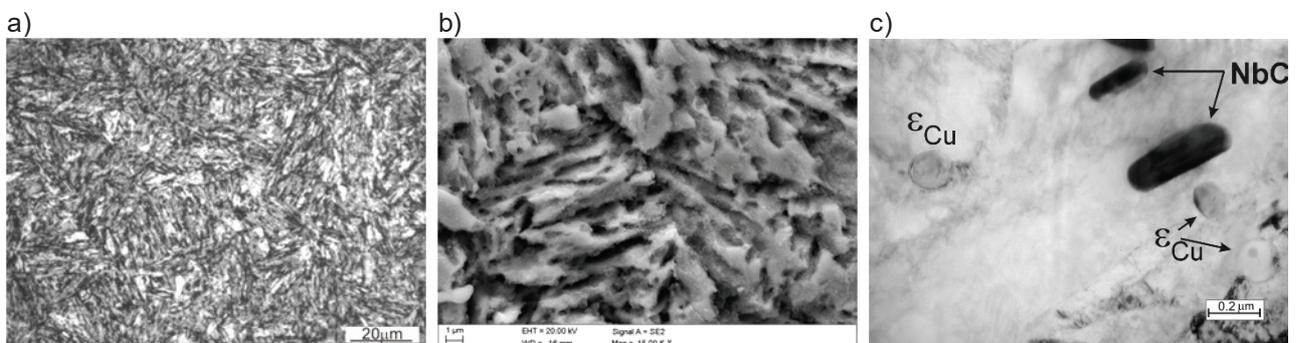


Fig. 7 The microstructure of investigated 17-4PH steel after quenching from 1040 °C and heated up to 600 °C with the rate of 0.5 °C/s: a) light microscope, b) SEM; c) TEM

Fig. 8 shows influence of temperature, up to which sample was heated after quenching (heating rate 0.05 °C/s), on hardness of investigated steel. The highest increase of hardness takes place in the temperature range of

450÷540 °C (in the range of precipitation of copper rich phase). Above 540 °C precipitation of the Cu_ε slows down and matrix recovery causes hardness decrease.

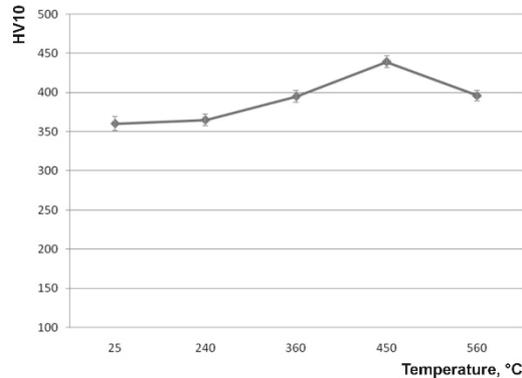


Fig. 8 Dependence of hardness of samples of investigated steel on the heating temperature after quenching (heating rate 0.05 °C/s)

4. CONCLUSIONS

In as-quenched state the microstructure of investigated steel consist of low carbon martensite, ferrite δ banding as well as of not dissolved during austenitizing niobium secondary carbides of MC type. Changes in the microstructure of the investigated steel during heating from the as-quenched state results from precipitation of transition iron carbides, transformation of small amount of retained austenite into martensite and mainly the Cu_ε phase precipitation. There were no significant changes in the chemical composition as well as in the morphology of the δ -ferrite.

During continuous heating from as-quenched state the highest hardness increase takes place in the temperature range of 450÷540 °C (in the range of precipitation of copper rich phase). Above 540 °C precipitation of Cu_ε slows down and matrix recovery causes hardness decrease.

ACKNOWLEDGEMENTS

A financial support of The National Centre for Research and Development Project No INNOLOT/II/9/NCBR/2013 "Advanced methods for manufacturing of aero engine case module components" is gratefully acknowledged.

REFERENCES

- [1] CHRISTIEN F., TELLING M.T.F., KNIGHT K.S. A comparison of dilatometry and in-situ neutron diffraction in tracking bulk phase transformations in a martensitic stainless steel. *Materials Characterization*, Vol. 82, 2013, pp. 50-57.
- [2] BAŁA, P., PACYNA, J., KRAWCZYK, J. The microstructure changes in high-speed steels during continuous heating from as-quenched state. *Metallic Materials*, vol. 49, 2011, pp. 125-130.
- [3] BAŁA, P., PACYNA, J. The kinetics of phase transformations during tempering of the new hot working tool steel designed for a large size forging dies. *Steel Research International Special edition Metal forming 2008*, Vol. 2, 2009, pp. 407-413.
- [4] BAŁA, P. The kinetics of phase transformations during tempering of tool steels with different carbon content. *Archives of Metallurgy and Materials*, Vol. 54, 2009, pp. 491-498.
- [5] BAŁA, P. Temporary process analysis based on the kinetics of phase transformations. *Archives of Metallurgy and Materials*, Vol. 54, 2009, pp. 1223-1230.

- [6] BALA, P. The dilatometric analysis of the high carbon alloys from Ni-Ta-Al-M system. *Archives of Metallurgy and Materials*, Vol. 59, 2014, pp. 977-980.
- [7] PAWŁOWSKI B., BAŁA P., DZIURKA R. Improper interpretation of dilatometric data for cooling transformation in steels. *Archives of Metallurgy and Materials*, Vol. 59, No.3, 2014, pp.1159-1161.
- [8] PAWŁOWSKI B. Determination of critical points of hypoeutectoid steel. *Archives of Metallurgy and Materials*, Vol. 57, No. 4, 2012, pp.957-962.
- [9] BALA P., Microstructure Characterization of High Carbon Alloy from the Ni-Ta-Al-Co-Cr System, *Archives of Metallurgy and Materials*. Vol. 57, 2012, pp. 937-941.
- [10] BALA P., Microstructural Characterization of the New Tool Ni-Based Alloy with High Carbon and Chromium Content, *Archives of Metallurgy and Materials*, Vol. 55, 2010, pp. 1053-1059.
- [11] CIOS G., BALA P., STEPIEN M., GORECKI K., Microstructure of Cast Ni-Cr-Al-C Alloy, *Archives of Metallurgy and Materials*, Vol. 60, 2015, pp. 145-148.
- [12] HSIAO C.N., CHIOU C.S., YANG J.R. Aging reactions in a 17-4 PH stainless steel. *Materials Chemistry and Physics*, Vol. 74, 2002, pp. 134-142.
- [13] NAKHAIE D, MOAYED M.H. Pitting corrosion of cold rolled solution treated 17-4 PH stainless steel. *Corrosion Science*, Vol. 80, 2014, pp. 290-298.
- [14] RACK H. J., KALISH D. The Strength, Fracture Toughness and Low Cycle Fatigue Behavior of 17 - 4 PH Stainless steel. *Metallurgical Transactions*, Vol. 5, 1974, pp. 1595-1605.
- [15] VISWANATHAN U. K., BANERJEE S., KRISHNAN R. Effects of Aging on the Microstructure of 17-4 PH Stainless Steel. *Materials Science and Engineering*, Vol. 104, 1988, pp. 181-189.